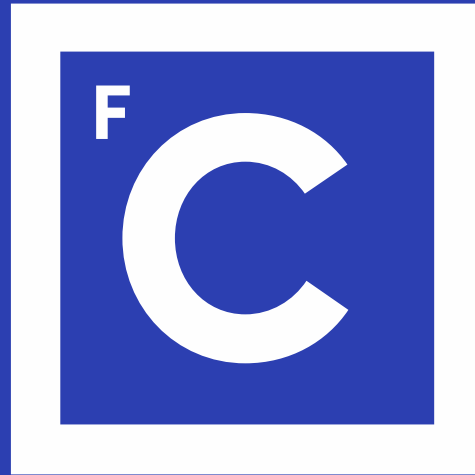




Ciências ULisboa

Faculdade
de Ciências
da Universidade
de Lisboa

DISCIPLINA MIEA 2018

move ► green



Mobilidade Sustentável

Home  Work/ University

on a regular basis

Commuting by motorized or soft modes



Emissions

Energy





Inter-terminal Gatwick airport 2008





Barcelona Metro line 9 without
train driver 2009



Heathrow Personal Rapid Transit system 2011

18 low-energy, driverless vehicles can each carry four passengers and their luggage.
(Ultra PRT | www.ultraprt.com.)

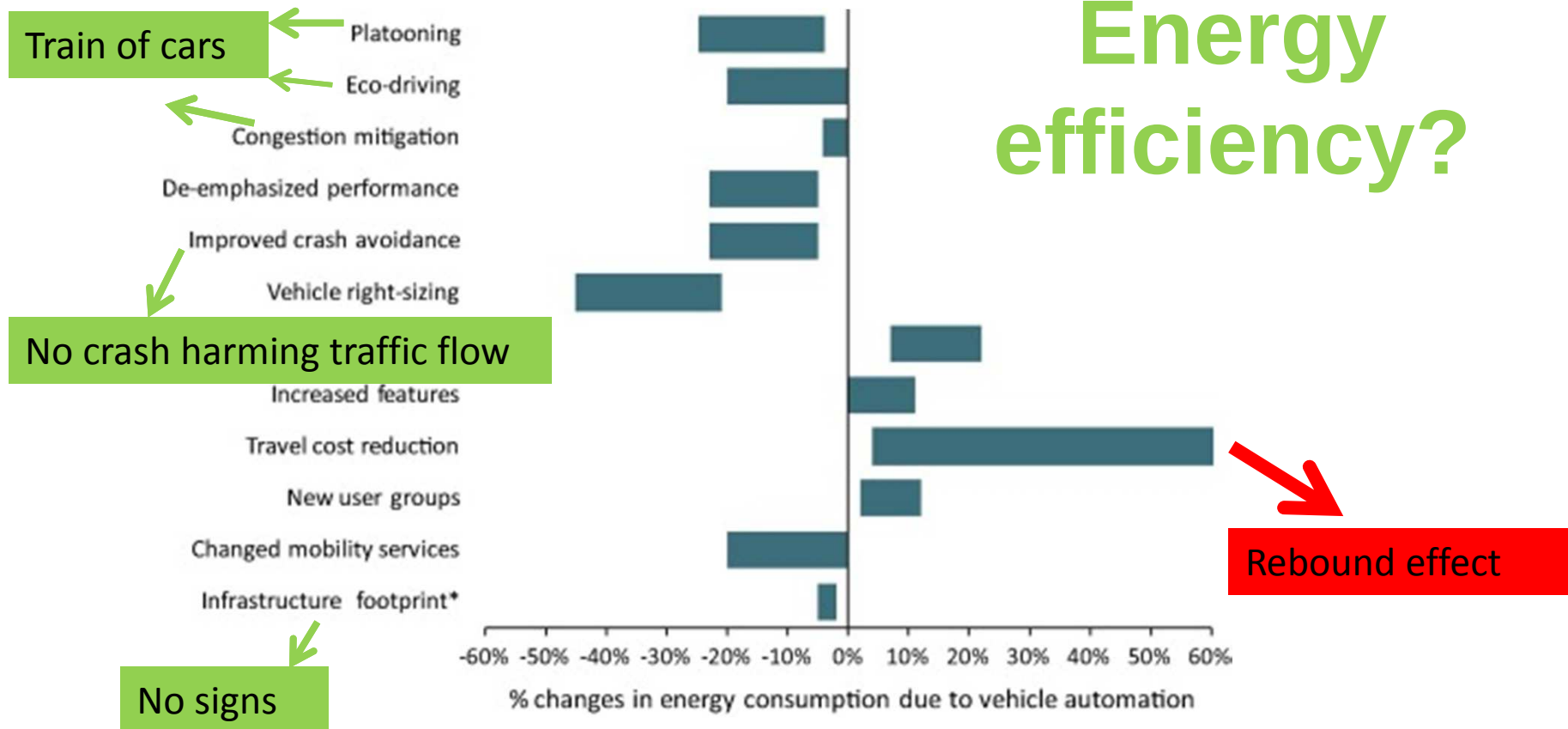




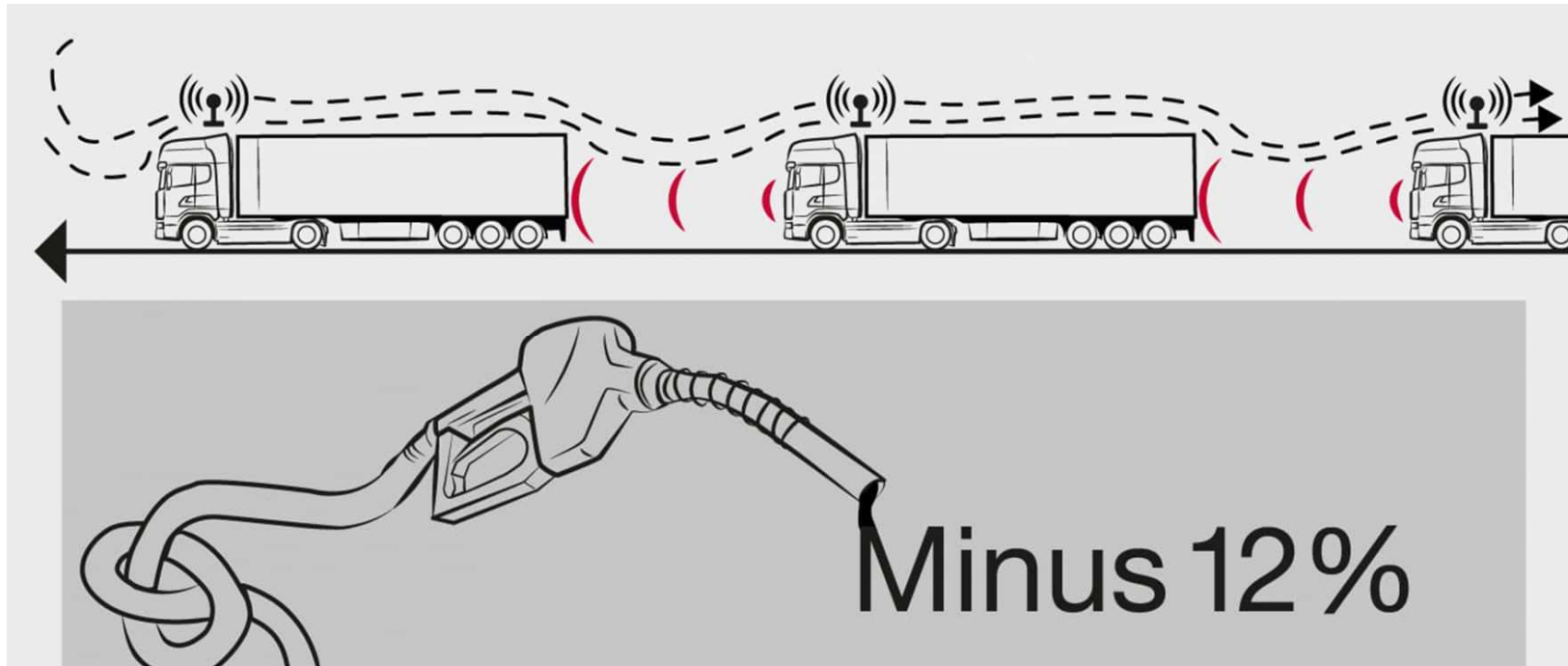
- **Autonomous vehicles**

Washinton D.C. 2016

Self-driving cars and energy consumption

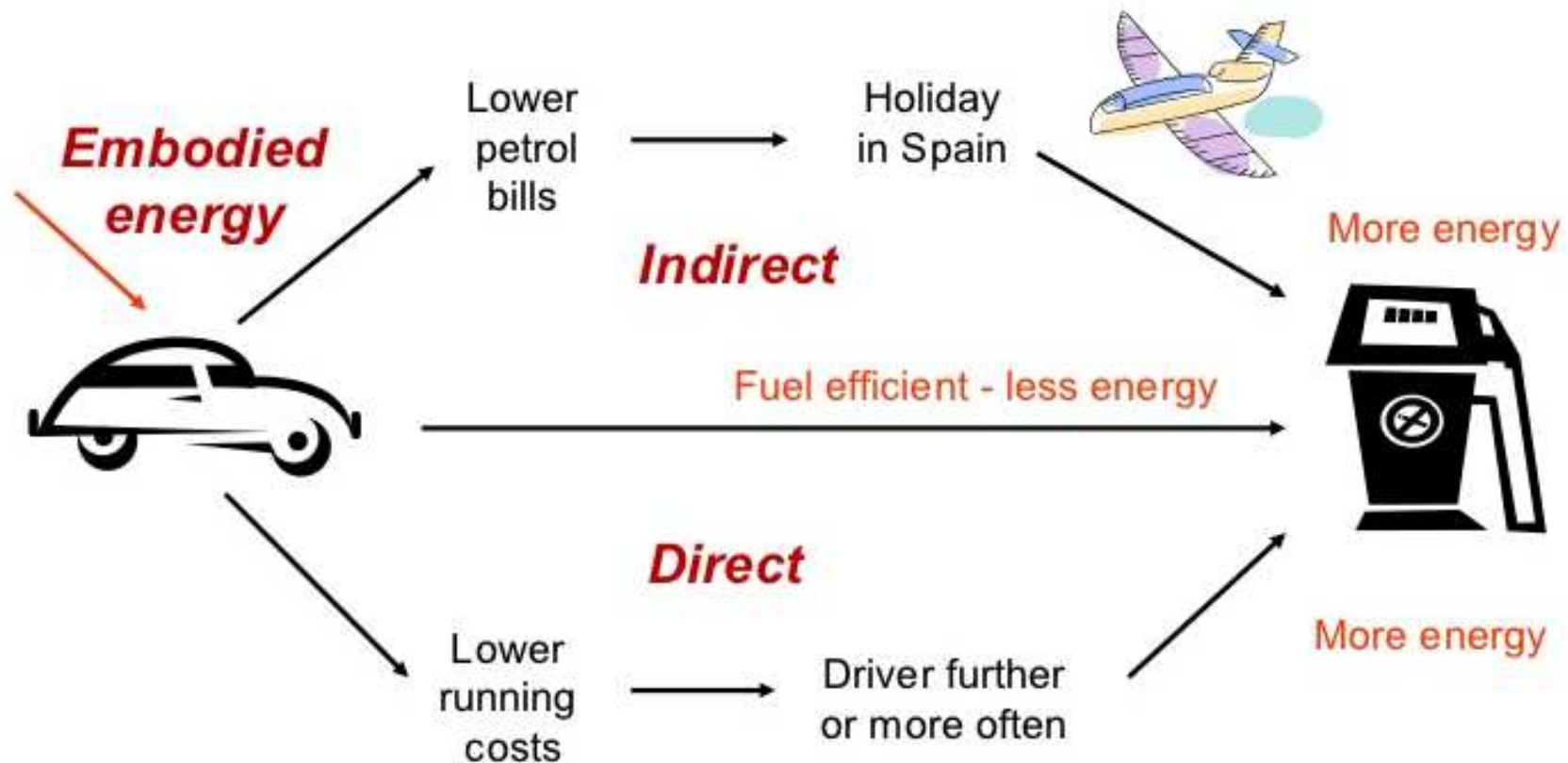


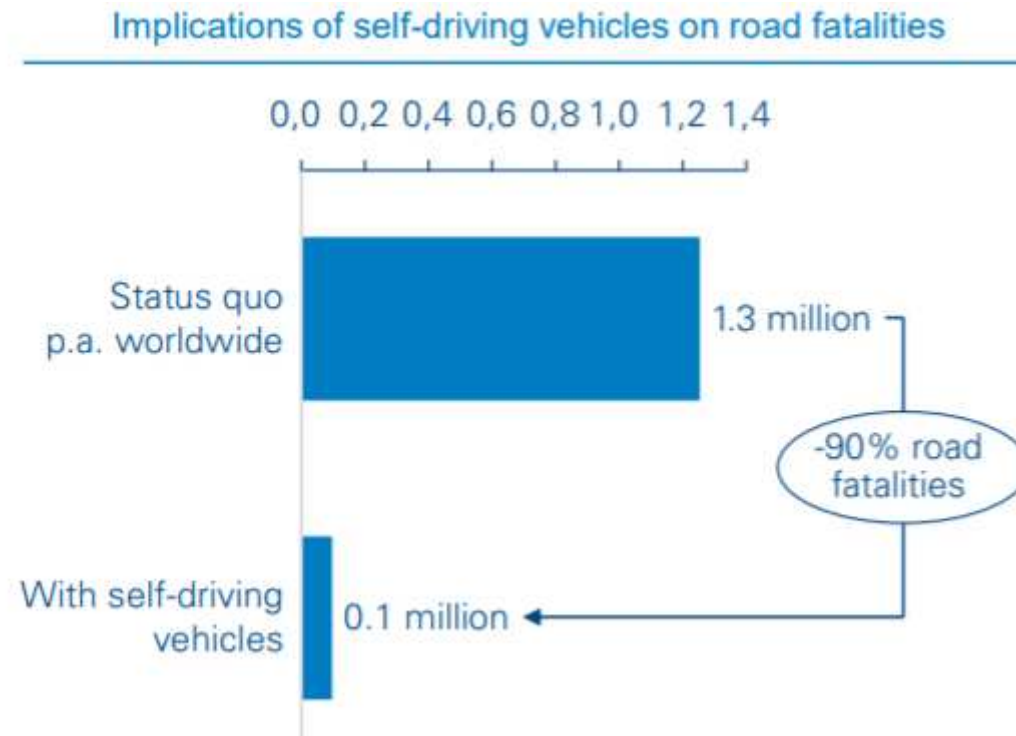
@ Help or hindrance? The travel, energy and carbon impact of highly automated vehicles Article · Apr 2016 · Transportation Research Part A Policy and Practice

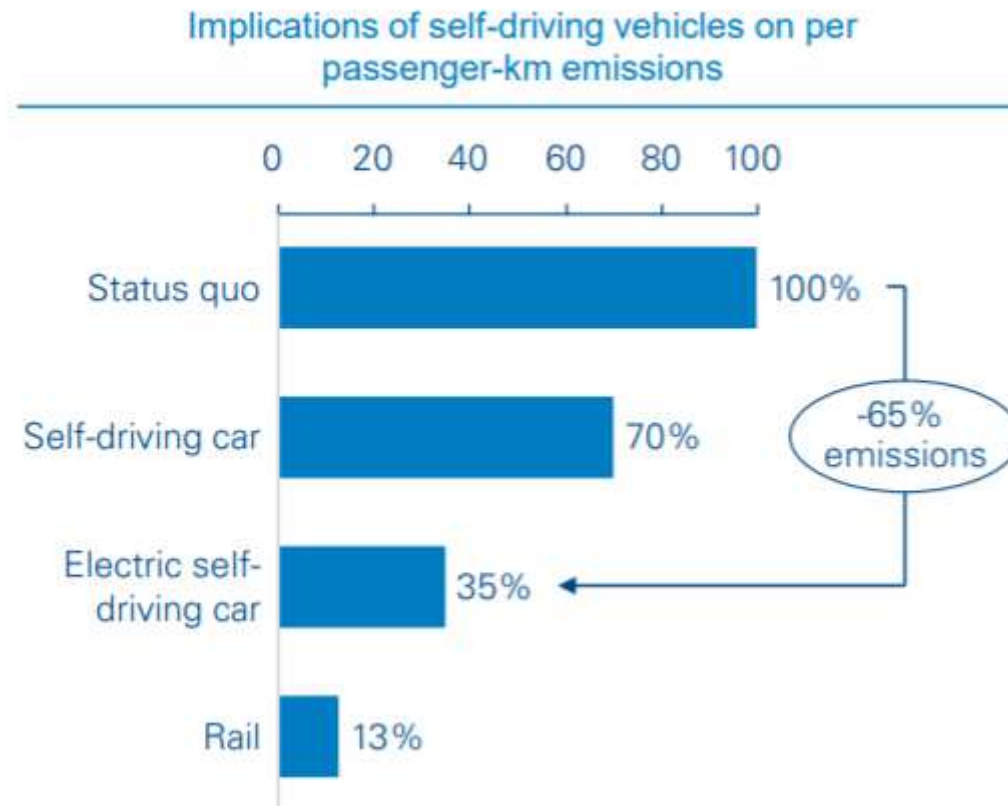


@ <https://www.scania.com/group/en/platooning-saves-up-to-12-percent-fuel/>

The rebound effect in plain terms is the misconception that using energy-efficient technology significantly cuts energy consumption and greenhouse gas emissions







Criteria	Weight	Definition
19 Transport-related CO ₂ emissions	4	- Ratio between the total amount of carbon dioxide emitted by the agglomeration area p.a. as a consequence of its transport activities and its population - The data considers carbon dioxide emissions from the burning of fossil fuels in transportation only (sectorial approach)
20 NO ₂ concentration	4	Annual arithmetic average of the daily concentrations of NO₂ recorded at all monitoring stations within the agglomeration area
21 PM ₁₀ concentration	4	- Annual mean concentration of particulate matter of less than 10 microns of diameter (PM₁₀) [ug/m³] in a city/ agglomeration area - For most of cities data from the WHO Global Urban Ambient Air Pollution Database was used (update 2016). Other sources if the city was not available in the WHO Database
22 PM _{2.5} concentration	4	- Annual mean concentration of particulate matter of less than 2.5 microns of diameter (PM_{2.5}) [ug/m³] in a city/ agglomeration area - For most cities data from the WHO Global Urban Ambient Air Pollution Database was used (update 2016). Other sources if the city was not available in the WHO Database

Minimum and maximum values of indicator scale (maximum 16 points out of 100 for CO₂, NO₂, PM₁₀ and PM_{2.5})

e.g. PM₁₀ 150 / 10 (µg/m³)

0 – 4

e.g. CO₂ kg/capita/year 7.5- 0

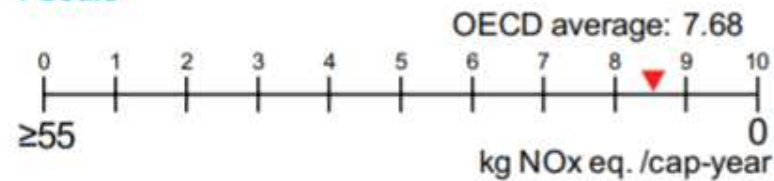


wbcasd

Pollutant	Relative importance (based on 2007USD) with NO _x , cost as reference
NO _x	1.00
PM ₁₀	1.06

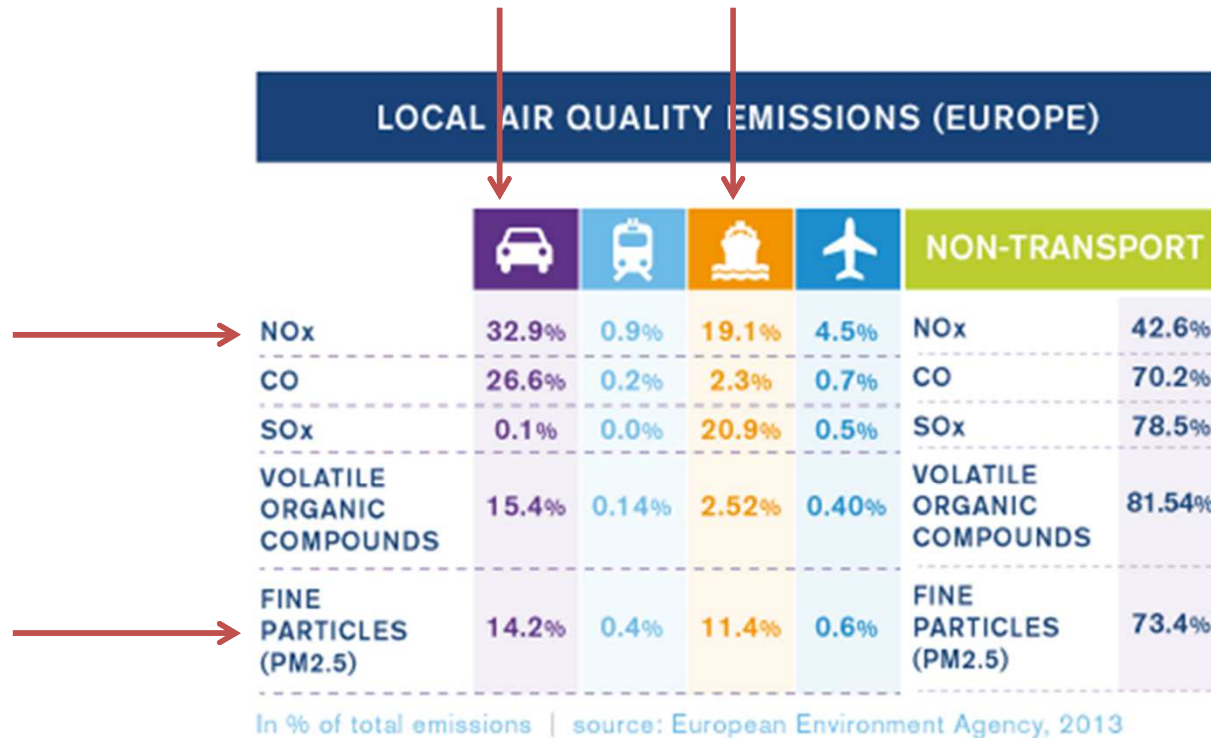
(Source: AEA Technology (2005) and Wang, Santini & Warinner (1994), US cities as in Victoria Transport Policy Institute (2011), www.vtpi.org).

f Scale

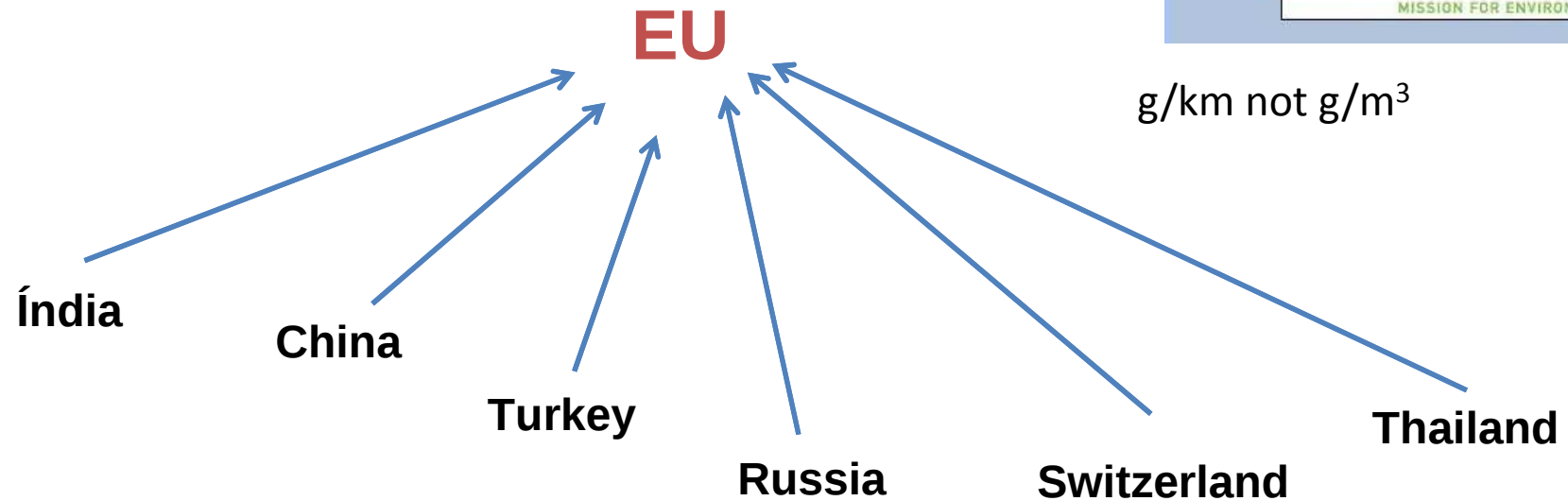
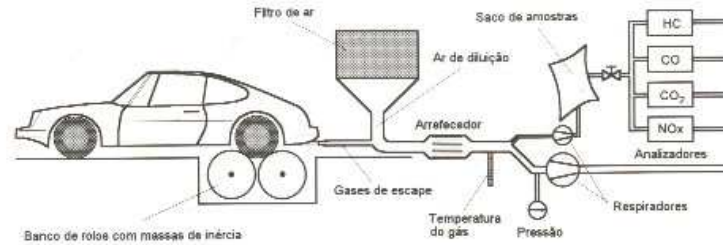


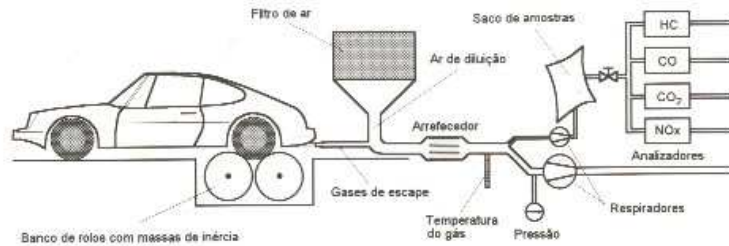
→ 0: ≥55 [kg NOx eq./ cap-year]

→ 10: 0 [kg NOx eq./ cap-year]



THANKS TO IMPROVEMENTS IN AIRCRAFT TECHNOLOGY,
THE IMPACT OF AIRCRAFT EMISSIONS ON LOCAL AIR QUALITY
IS RELATIVELY LOW COMPARED TO OTHER SOURCES.





g/km

Premium

CERC - Version ADMS-Urban - Air Pollution Modelling Software

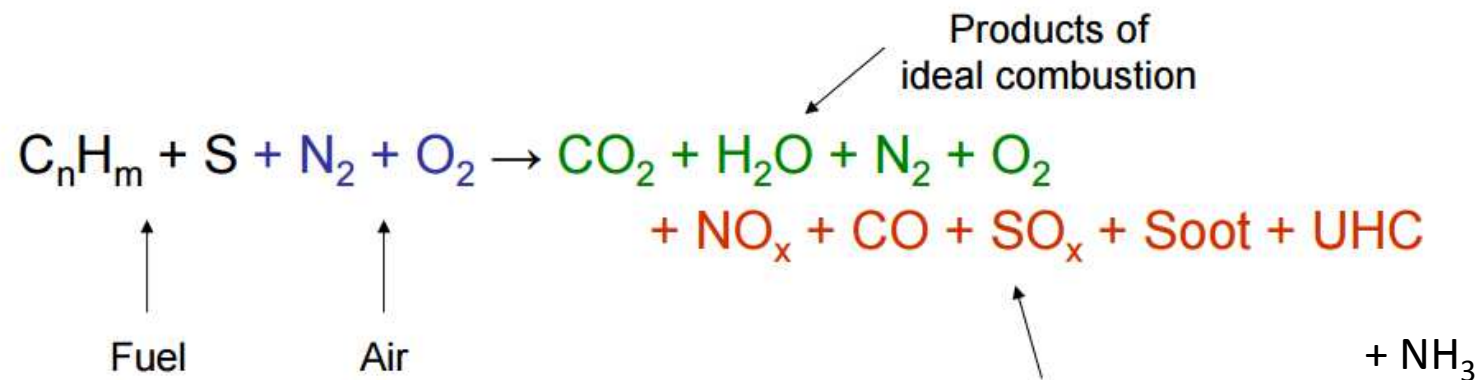


ADMS-Urban is a comprehensive system for modelling air quality in large urban areas, cities and towns. It is the only practical urban air quality model which incorporates the latest scientific understanding, explicitly represents the full range of source types occurring in an urban area, takes account of complex urban morphology including ...

By Cambridge
Environmental Research
Consultants (CERC) based
in **Cambridge, UNITED
KINGDOM.**
from Air Pollution
Modelling Software
Software line

g/m³

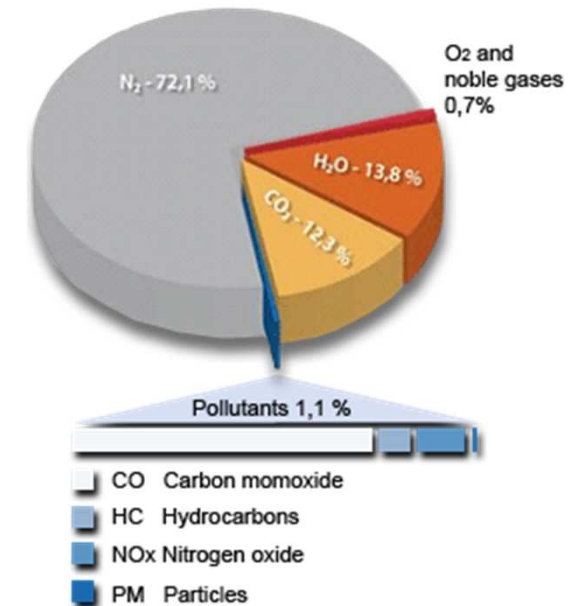
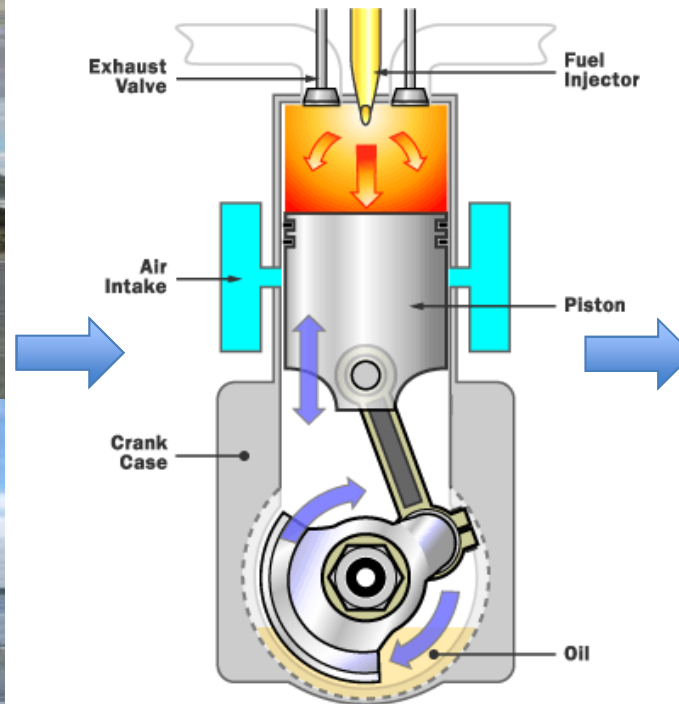
Why do we have emissions???



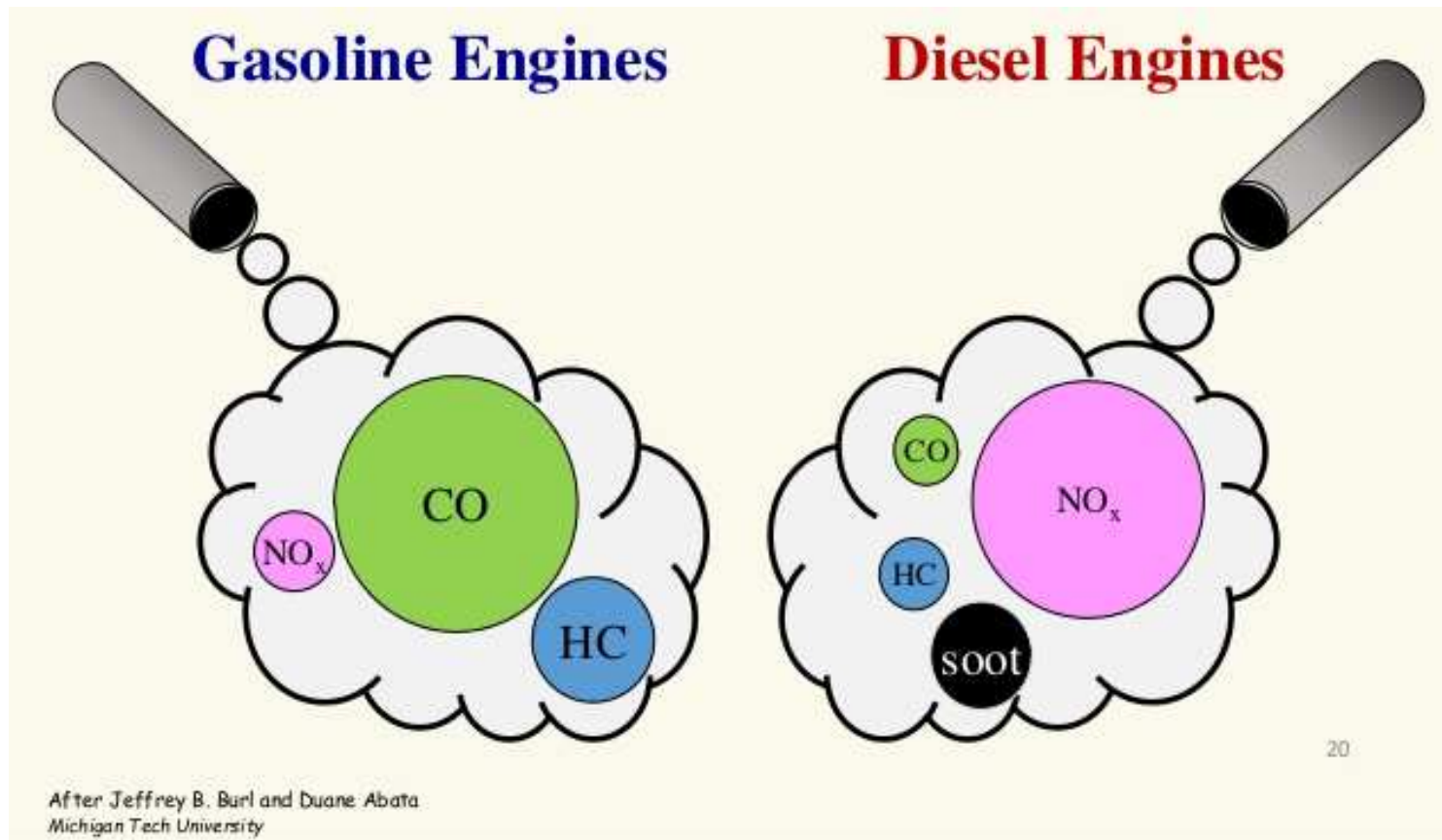
NO_x: Affects ozone (O₃) concentration
 CO₂: Absorbs outgoing infrared radiation
 CO: Toxic
 Soot: Visible

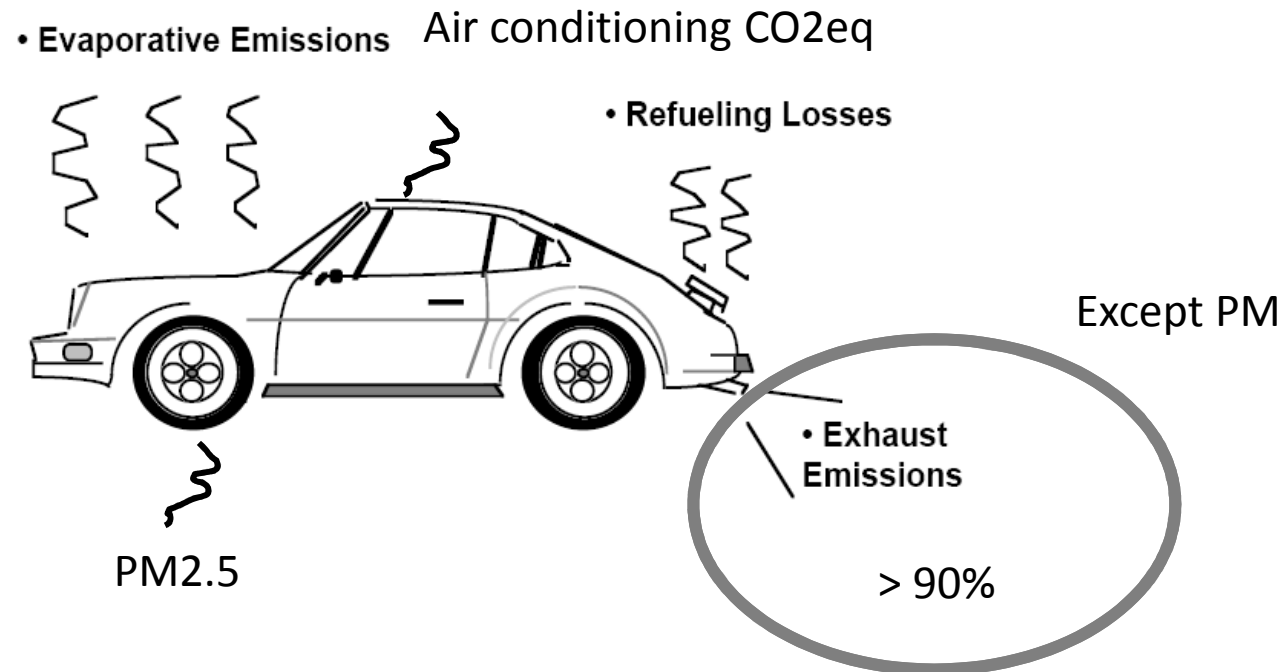


Why do we have emissions???



MOBILE COMBUSTION





MOBILE COMBUSTION

Global

CO₂

CO₂, N₂O, CH₄

• Evaporative Emissions

• Refueling Losses

• Exhaust Emissions

Tailpipe

CH₄, N₂O ~ 0.3-3% of tailpipe CO₂
TABLE 1.8

Direct Greenhouse Gas **Emissions** from Passenger Cars on Petrol, Diesel, LPG and CNG under Real-World Driving Conditions

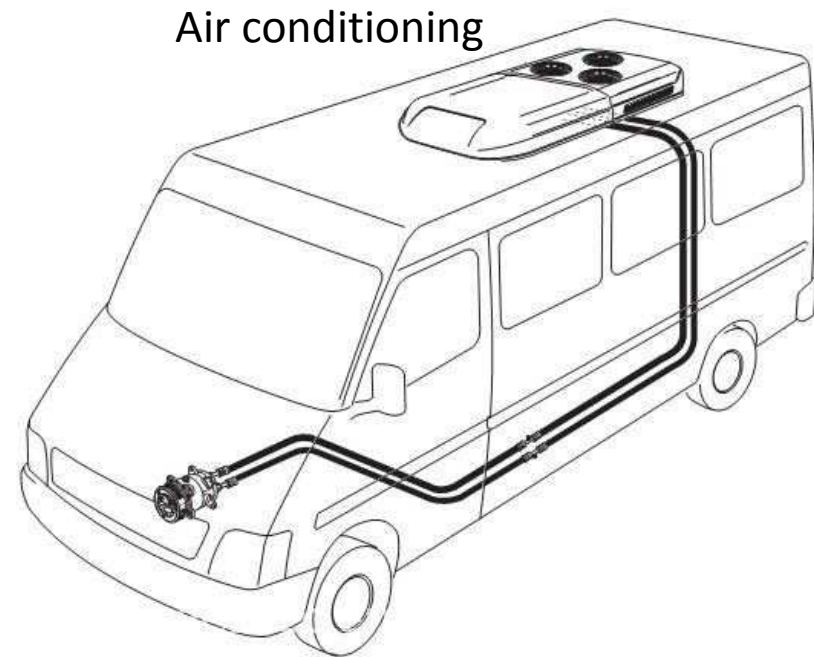
Fuel	CO ₂ (g/km)	CH ₄ (g/km)	N ₂ O (g/km)	GHG Emission (g of CO ₂ Equivalent/km)
Petrol	208.1	0.009	0.003	209.2
Diesel	180.5	0.004	0.007	182.7
LPG	189.3	0.007	0.003	190.4
CNG	168.6	0.0074	0.001	170.6

Source: Adapted from P. Hendriksen et al., Evaluation of the environmental performance of modern passenger cars running on petrol, diesel, automotive LPG and CNG, TNO-report 03.OR.VM.055.1/PHE, TNO Automotive, December 2003.

Global

CO₂ equivalent

Refrigerante	Fórmula química	CO ₂ eq 100 anos
HFC-134a	CH ₂ FCF ₃	1430
SF ₆	SF ₆	22800
PFC	CF ₄	7390



Refrigerant : R-134a.
 Refrigerant charge : 1800 ± 100 gr.
 Oil : SP-20.
 Oil charge : 110 cc (In addition to the existing oil in the compressor).

HFC-134a ~ 3-5% of tailpipe CO₂ or 6-8 g/km of CO₂ eq

CO₂ equivalent

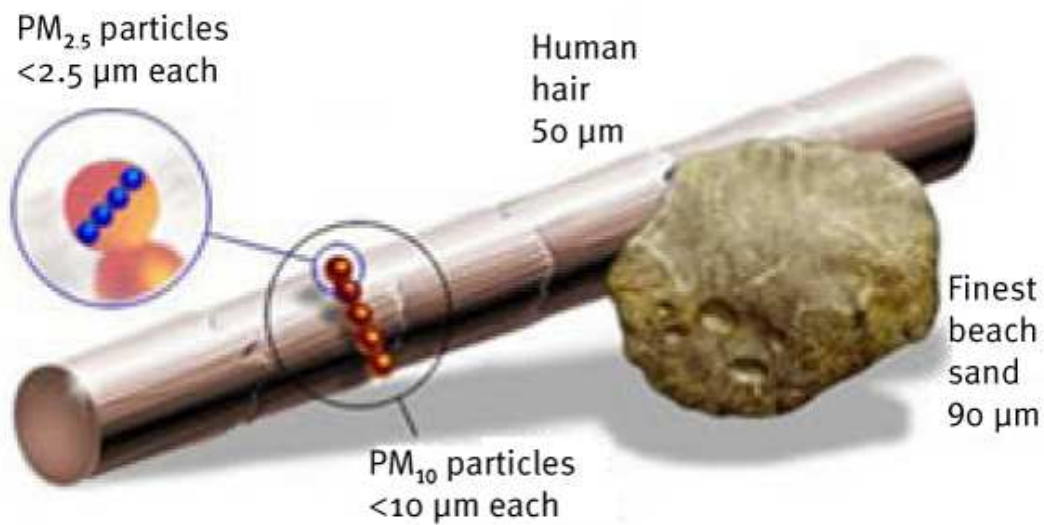
$$\text{CO}_2 \text{ eq [kg]} = \sum_i^n (\text{GWP}_{100} * \text{Emission}_i \text{ [kg]})$$

Intergovernmental Panel on Climate Change (IPCC)

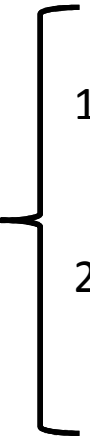


Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m ⁻² ppb ⁻¹)	Global Warming Potential for Given Time Horizon			
				SAR ¹ (100-yr)	20-yr	100-yr	500-yr
Carbon dioxide	CO ₂	See below ^a	^b 1.4x10 ⁻⁵	1	1	1	1
Methane ^c	CH ₄	12 ^c	3.7x10 ⁻⁴	21	72	25	7.6
Nitrous oxide	N ₂ O	114	3.03x10 ⁻³	310	289	298	153
Substances controlled by the Montreal Protocol							
CFC-11	CCl ₃ F	45	0.25	3,800	6,730	4,750	1,620
CFC-12	CCl ₂ F ₂	100	0.32	8,100	11,000	10,900	5,200
CFC-13	CClF ₃	640	0.25		10,800	14,400	16,400
CFC-113	CCl ₂ FCF ₂	85	0.3	4,800	6,540	6,130	2,700
CFC-114	CClF ₂ CClF ₂	300	0.31		8,040	10,000	8,730
CFC-115	CClF ₂ CF ₃	1,700	0.18		5,310	7,370	9,990
Halon-1301	CBrF ₃	65	0.32	5,400	8,480	7,140	2,760
Halon-1211	CBrClF ₂	16	0.3		4,750	1,890	575
Halon-2402	CBrF ₂ CBrF ₂	20	0.33		3,680	1,640	503
Carbon tetrachloride	CCl ₄	26	0.13	1,400	2,700	1,400	435

PM_{2.5} and PM₁₀



primary emissions of PM_{2.5}

- 
- 1) it is directly emitted from the tailpipes of cars, trucks and other on-road vehicles;
 - 2) it is re-entrained from materials found on the roadway (typically known as fugitive dust), and
 - 3) it is created by secondary formation from precursor emissions such as sulfur dioxide (SO₂), nitrogen oxides (NO_x), volatile organic compounds (VOCs) and ammonia (NH₃).

30% to 90% of all PM_{2.5}

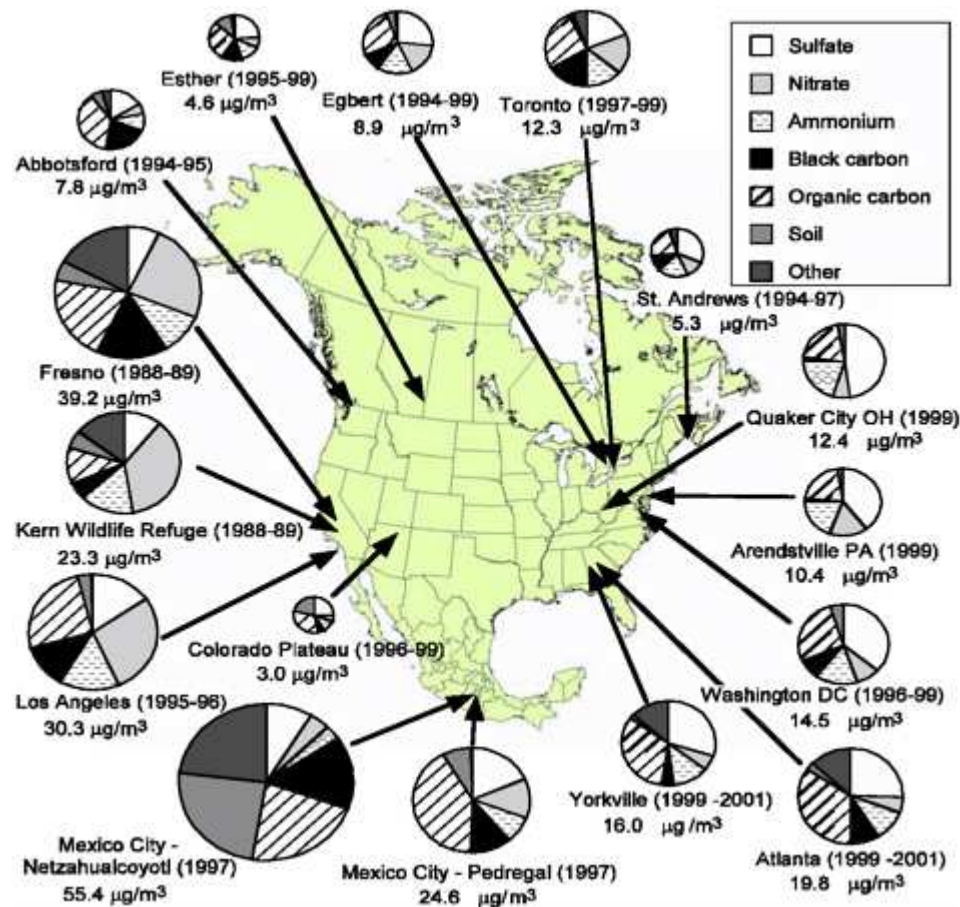
Source: <https://www3.epa.gov/ttnchie1/conference/ei13/mobile/hodan.pdf>

2)

- paved roads contributes between 3% and 16% of total directly emitted PM2.5, and
- unpaved roads contribute between 9% and 22% of total directly emitted PM2.5



PM2.5 and PM10



Minerals (Fe, Ca, Si)

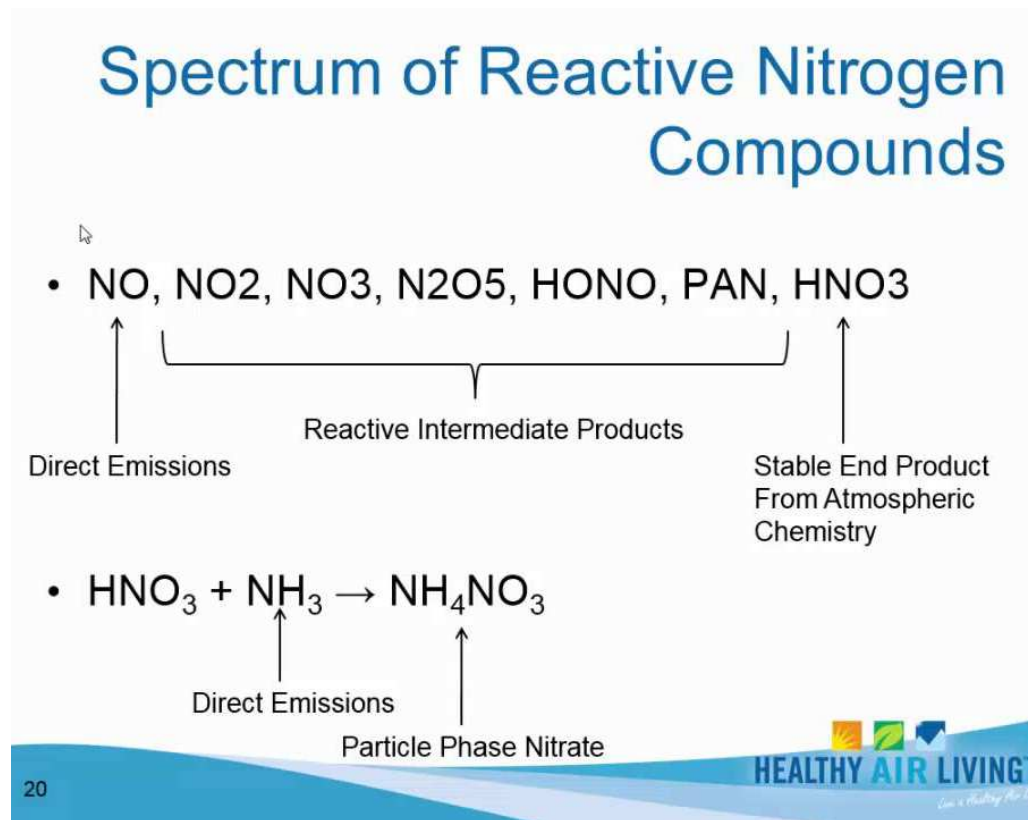
Sulfates (SO_4^{2-})

Nitrates (NH_4^+)

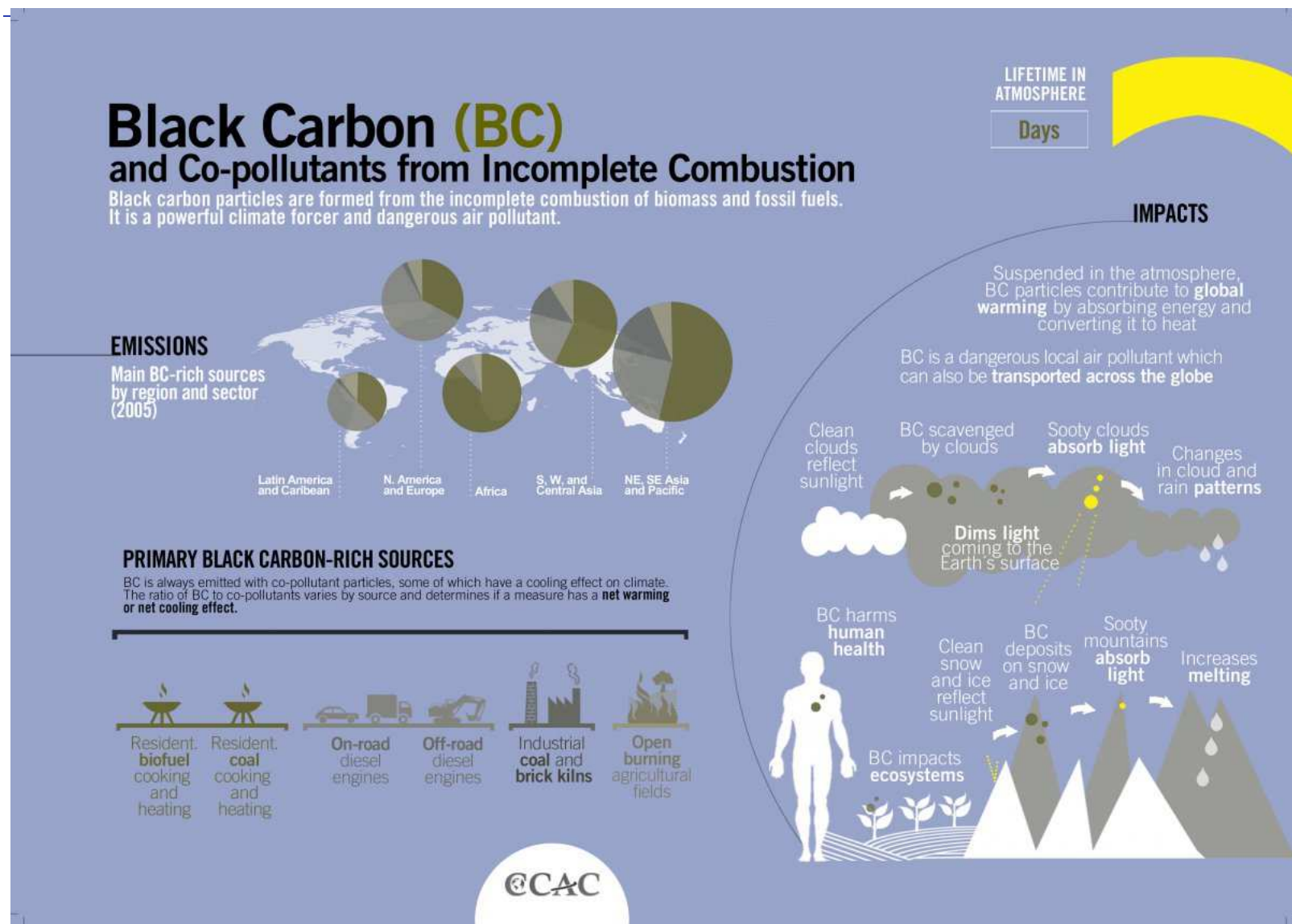
Organic carbon

Elemental carbon

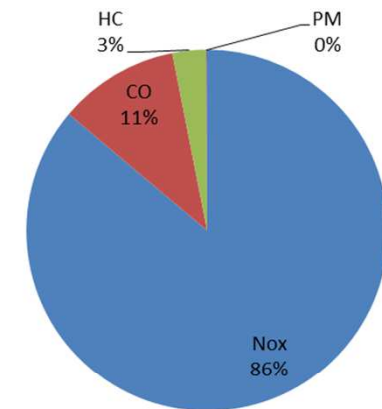
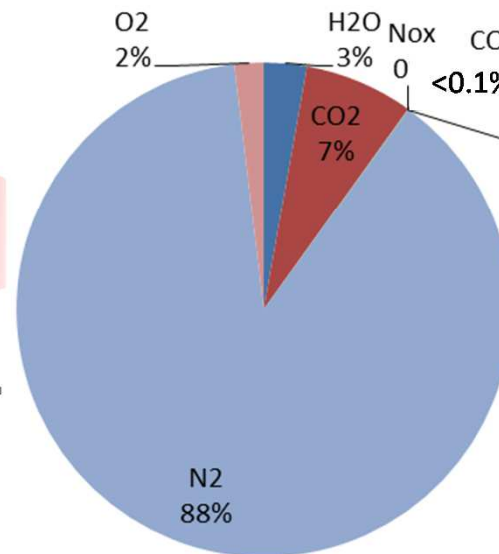
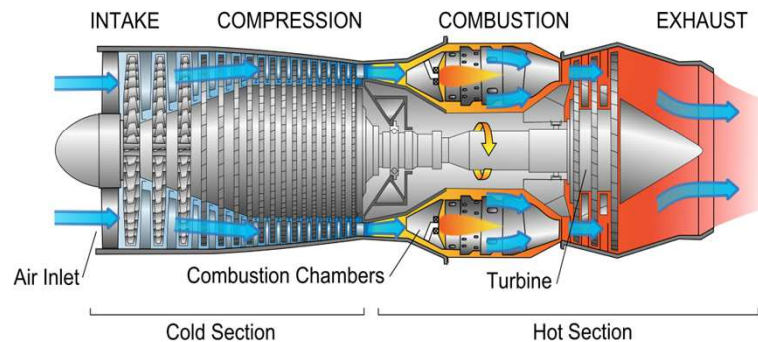
PM2.5 and PM10



Nitrates (NH₄⁺)



Why do we have emissions???

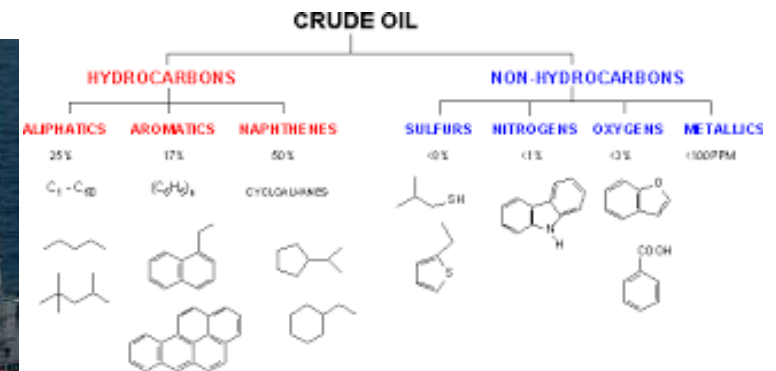


MOBILE COMBUSTION

1 toe = 41.868 GJ or 11.63 MWh

the amount of energy released by burning one tonne of crude oil

1 toe ~ 2.5 tCO₂



Carbon - 83 to 87%

Hydrogen - 10 to 14%

Nitrogen - 0.1 to 2%

Oxygen - 0.05 to 1.5%

Sulfur - 0.05 to 6.0%

Metals - < 0.1%



		Density at STP (kg/m ³)	Ratio of HHV to LHV energy content	Net Calorific Value / LHV (MJ/L) (MJ/kg)		Gross Calorific Value / HHV (MJ/L) (MJ/kg)		Carbon Intensity (g CO ₂ -eq / MJ LHV)
Crude Oil		856 ± 24	1.052 ± 0.001	36.84 ± 1.05	43.05 ± 1.40	38.76 ± 1.10	45.30 ± 1.47	73.5 ± 2.6
Petrol / Gasoline		741 ± 4	1.063 ± 0.015	32.70 ± 0.44	44.15 ± 0.74	34.77 ± 0.47	46.94 ± 0.70	70.8 ± 4.4
Diesel		837 ± 8	1.063 ± 0.011	35.94 ± 0.45	42.91 ± 0.46	38.19 ± 0.47	45.60 ± 0.49	74.3 ± 2.3
Fuel Oil		959 ± 17	1.058 ± 0.008	39.21 ± 1.09	40.87 ± 0.94	41.50 ± 1.15	43.26 ± 1.00	77.8 ± 2.1
LPG		533 ± 18	1.077 ± 0.008	24.67 ± 0.80	46.28 ± 0.74	26.57 ± 0.86	49.84 ± 0.80	63.9 ± 2.1
Kerosene		807 ± 6	1.053 ± 0.001	35.24 ± 0.41	43.69 ± 0.51	37.10 ± 0.43	45.99 ± 0.54	72.0 ± 1.8
Hydrogen	(35 MPa)	23.65 ± 0.09		2.837 ± 0.003		3.355 ± 0.004		
	(70 MPa)	39.69 ± 0.16	1.183 ± 0.001	4.761 ± 0.005	119.95 ± 0.13	5.631 ± 0.006	141.88 ± 0.16	0
	(liquid)	72.41 ± 0.72		8.685 ± 0.010		10.273 ± 0.011		

		(kg/m ³)	(HHV / LHV)	(MJ/kg)		(MJ/kg)	(g/MJ LHV)	
Coal			1.050 ± 0.004	-	25.75 ± 2.64	-	27.05 ± 2.77	95.7 ± 7.0

		(kg/m ³)	(HHV / LHV)	(MJ/m ³)	(MJ/kg)	(MJ/m ³)	(MJ/kg)	(g/MJ LHV)
Natural Gas		0.768 ± 0.039	1.109 ± 0.003	35.22 ± 2.22	45.86 ± 3.95	39.05 ± 2.47	50.84 ± 4.38	56.9 ± 3.4
Hydrogen	(1 atm.)	0.0838 ± 0.0008	1.183 ± 0.001	10.05 ± 0.01	119.95 ± 0.13	11.88 ± 0.01	141.88 ± 0.16	0

Local

HC, CO, NOx and PM



Global

CO₂ equivalent or GHG





Local



4.2 million

deaths every year as a result of exposure to ambient (outdoor) air pollution

3.8 million

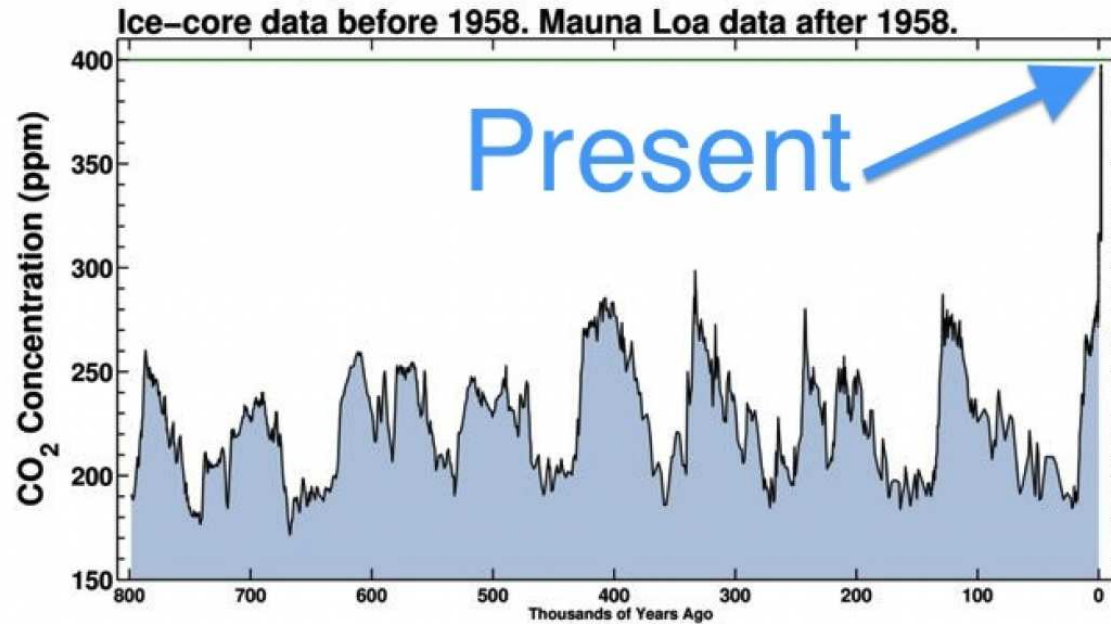
deaths every year as a result of household exposure to smoke from dirty cookstoves and fuels

91%

of the world's population lives in places where air quality exceeds WHO guideline limits



World Health
Organization

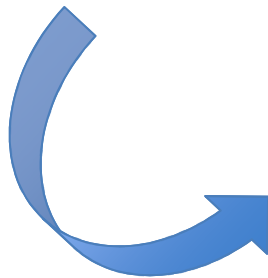


400 ppm CO₂ in outdoor environment

Global
CO₂ equivalent or GHG



400 ppm CO₂ in outdoor environment

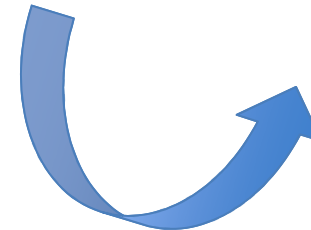


$$pV=nRT$$

PTN = Standard pressure and Temperature

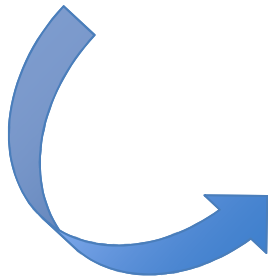
Pressure = 100 kPa

Temperature = 298 K

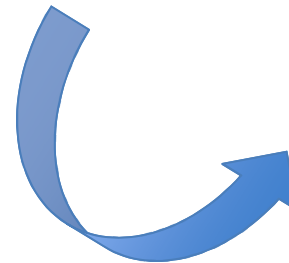


0.7 g/m³ CO₂ in outdoor environment

400 ppm CO₂ in outdoor environment

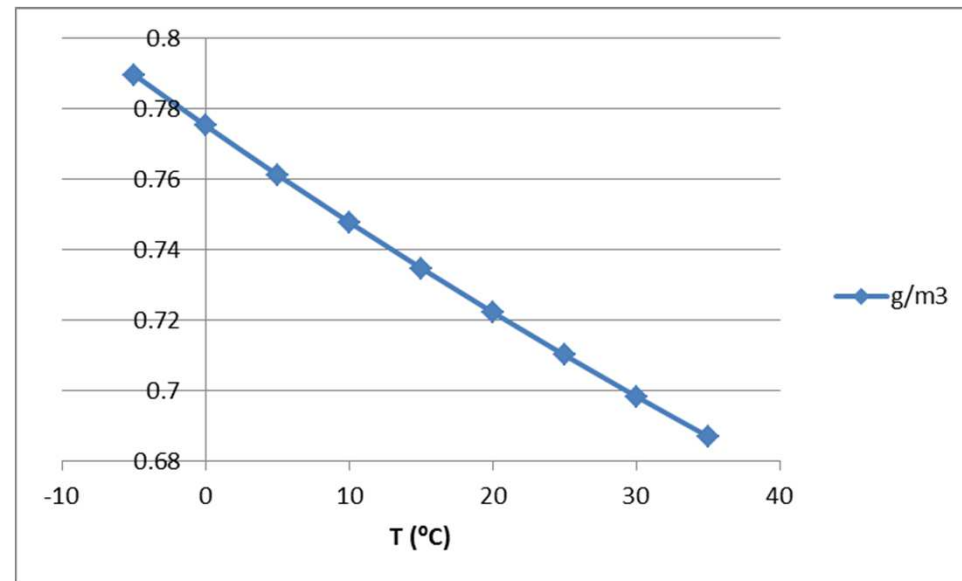


$$\frac{400 * 10^{-6} \text{ kmol CO}_2 * (12 + 2 * 16) \text{ kg/kmol CO}_2}{1 \text{ kmol air} \cdot \frac{8.314 \frac{\text{kJ}}{\text{kmol air} \cdot \text{K}} * 298 \text{ K}}{100 \text{ kPa}}}$$



0.7 g/m³ CO₂ in outdoor environment

Effect of temperature in real conditions.....

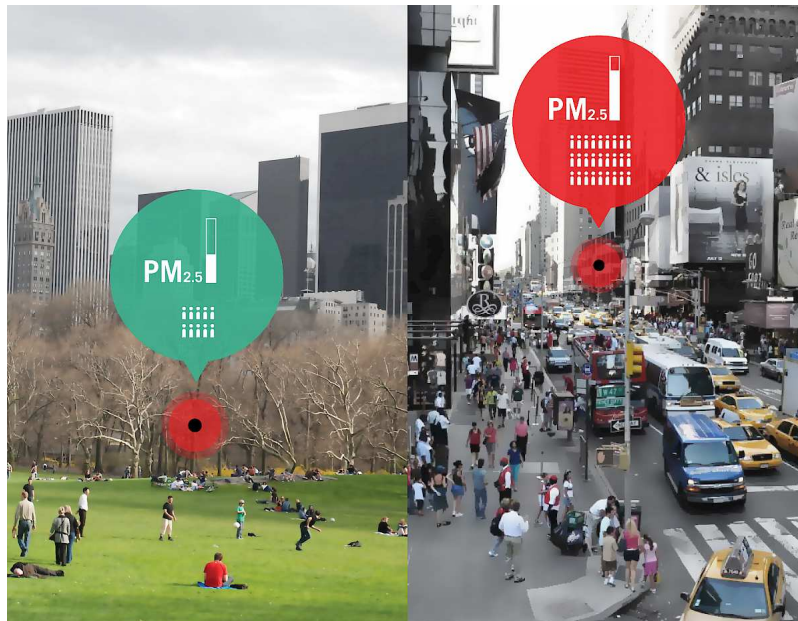


Partícle (PM)

PM₁₀ daily average $\leq 50 \mu\text{g}/\text{m}^3$

PM₁₀ anual average $\leq 40 \mu\text{g}/\text{m}^3$.

PM_{2.5} annual average $\leq 25 \mu\text{g}/\text{m}^3$.



Óxidos de azoto (NO_x) $< 30 \mu\text{g}/\text{m}^3$

Média anual

Óxidos de azoto (NO₂) $< 200 \mu\text{g}/\text{m}^3$

1 hora

Monóxido de carbono (CO) $< 10 \text{mg}/\text{m}^3$

Média máxima por períodos de 8 horas

Benzeno (C₆H₆) $< 5 \mu\text{g}/\text{m}^3$

Média anual

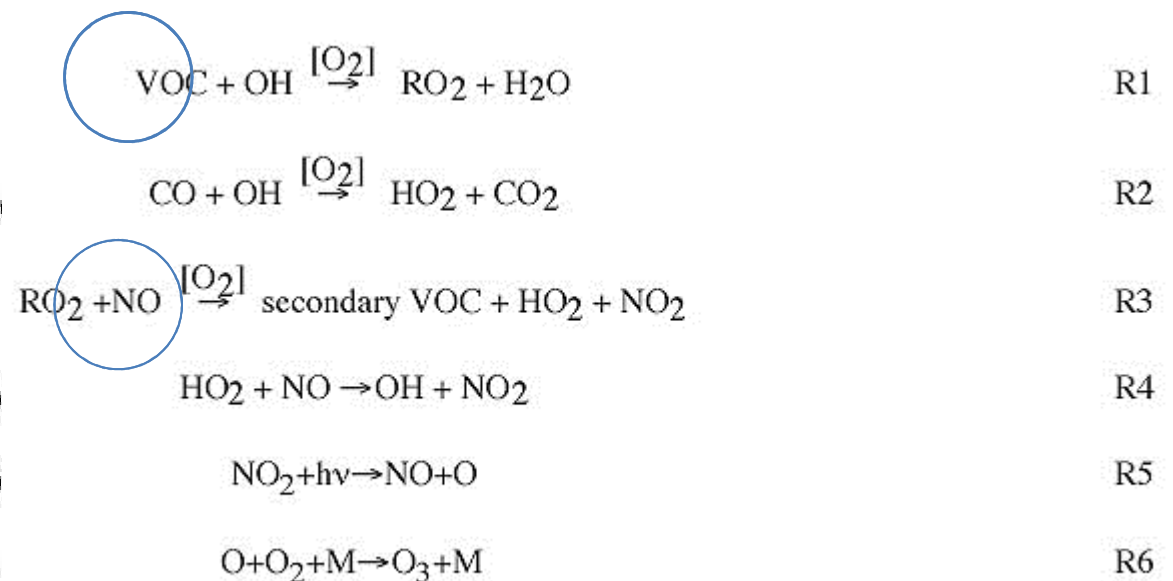
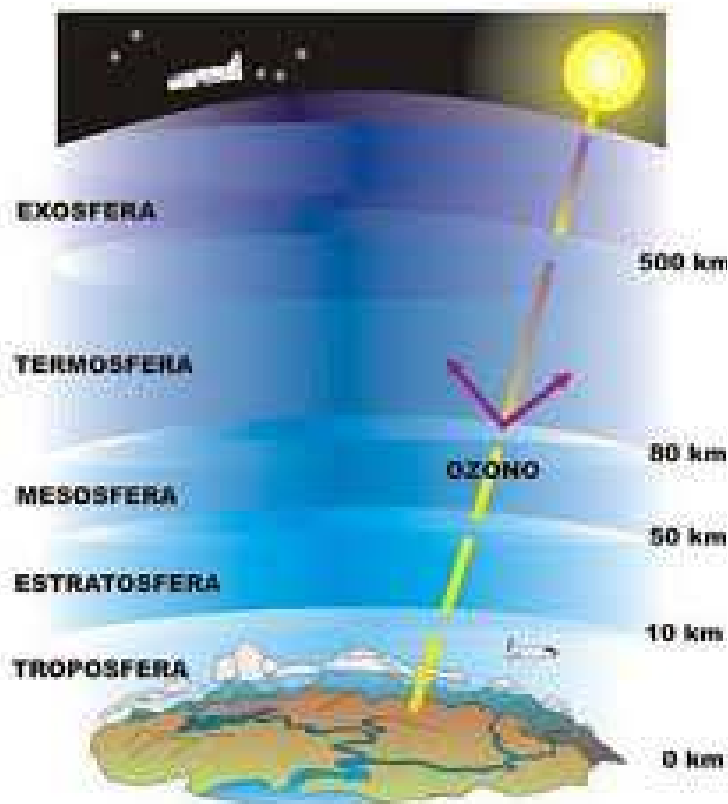
Dióxido de Enxofre (SO₂) $< 500 \mu\text{g}/\text{m}^3$

1 hora

Ozono (O₃) $< 240 \mu\text{g}/\text{m}^3$

1 hora

Tropospheric ozone O3 and smog Overview: Tropospheric ozone, smog and ozone-NOx-VOC sensitivity. Dr. Sanford Sillman, Research Scientist, University of Michigan



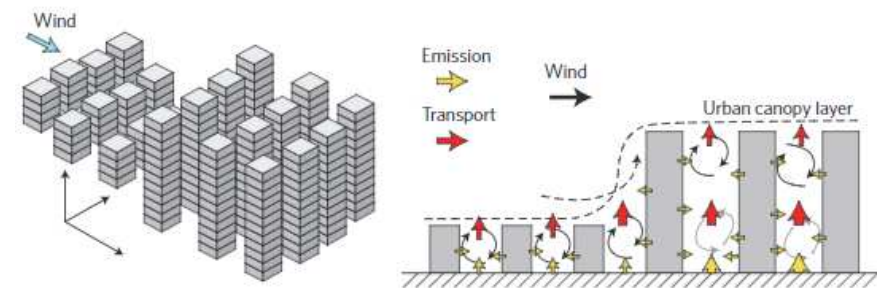
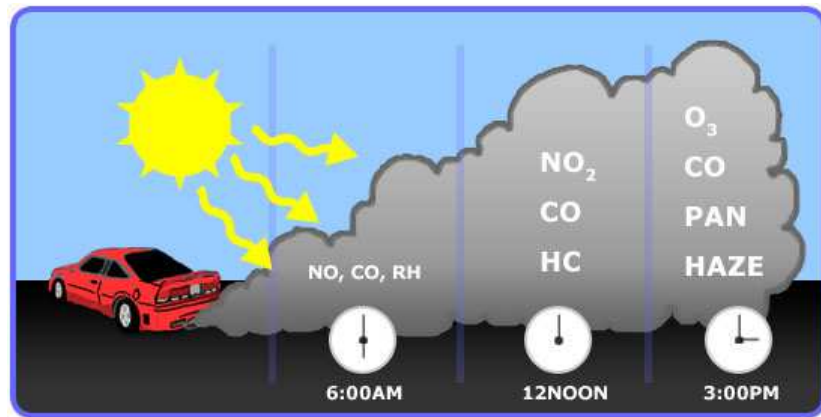
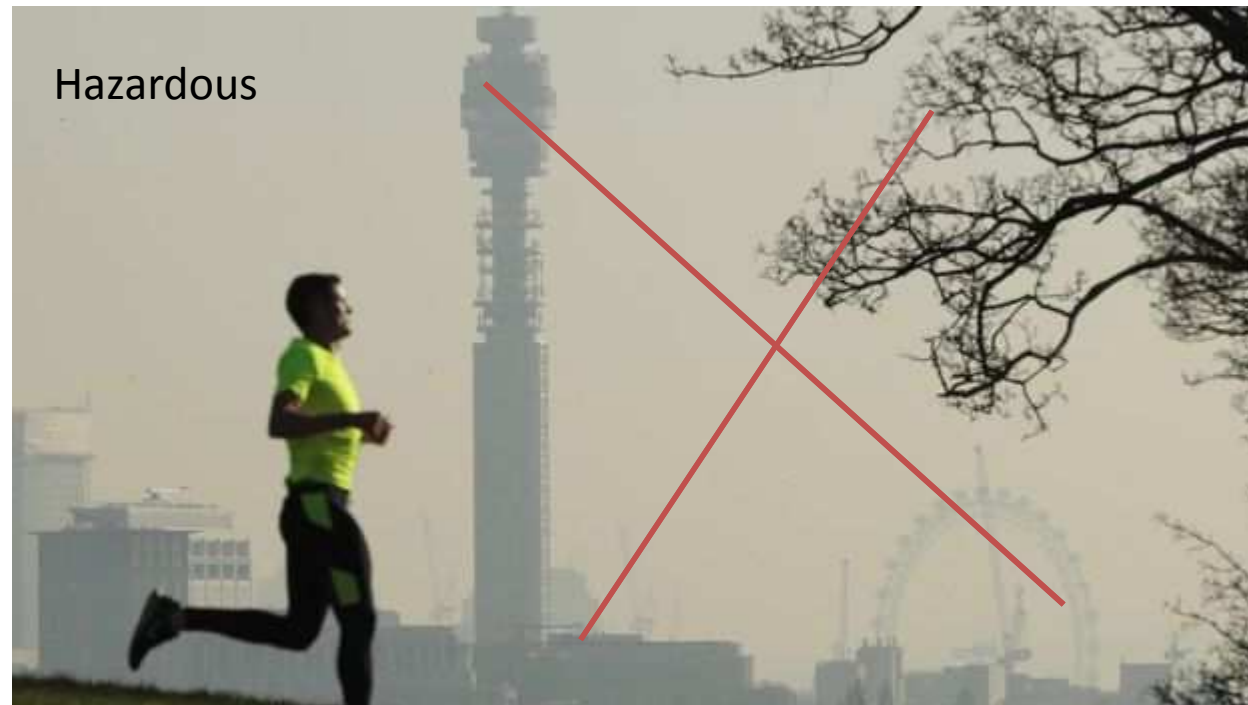


Figure 1 | Schematic illustration of air flow, pollutant concentration and dispersion with different building heights.

@ Air quality by urban design de Y Zhang - 2013

MONITORING NEEDED TO ADVISE PEOPLE!

e.g. colour scheme



Hazardous

O ₃ (ppb)	O ₃ (ppb)	PM _{2.5} (µg/m ³)	PM ₁₀ (µg/m ³)	CO (ppm)	SO ₂ (ppb)	NO ₂ (ppb)	AQI	AQI
$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$C_{low} - C_{high}(avg)$	$I_{low} - I_{high}$	Category
0-54 (8-hr)	-	0.0-12.0 (24-hr)	0-54 (24-hr)	0.0-4.4 (8-hr)	0-35 (1-hr)	0-53 (1-hr)	0-50	Good
55-70 (8-hr)	-	12.1-35.4 (24-hr)	55-154 (24-hr)	4.5-9.4 (8-hr)	36-75 (1-hr)	54-100 (1-hr)	51-100	Moderate
71-85 (8-hr)	125-164 (1-hr)	35.5-55.4 (24-hr)	155-254 (24-hr)	9.5-12.4 (8-hr)	76-185 (1-hr)	101-360 (1-hr)	101-150	Unhealthy for Sensitive Groups
86-105 (8-hr)	165-204 (1-hr)	55.5-150.4 (24-hr)	255-354 (24-hr)	12.5-15.4 (8-hr)	186-304 (1-hr)	361-649 (1-hr)	151-200	Unhealthy
106-200 (8-hr)	205-404 (1-hr)	150.5-250.4 (24-hr)	355-424 (24-hr)	15.5-30.4 (8-hr)	305-604 (24-hr)	650-1249 (1-hr)	201-300	Very Unhealthy
-	405-504 (1-hr)	250.5-350.4 (24-hr)	425-504 (24-hr)	30.5-40.4 (8-hr)	605-804 (24-hr)	1250-1649 (1-hr)	301-400	Hazardous
-	505-604 (1-hr)	350.5-500.4 (24-hr)	505-604 (24-hr)	40.5-50.4 (8-hr)	805-1004 (24-hr)	1650-2049 (1-hr)	401-500	

USEPA AQI (<https://waqi.info/>)



CAQI (<http://airindex.eea.europa.eu/>)



Pollutant	Index level				
	(based on pollutant concentrations in µg/m ³)				
	Good	Fair	Moderate	Poor	Very poor
Particles less than 2.5 µm (PM _{2.5})	0-10	10-20	20-25	25-50	50-800
Particles less than 10 µm (PM ₁₀)	0-20	20-35	35-50	50-100	100-1200
Nitrogen dioxide (NO ₂)	0-40	40-100	100-200	200-400	400-1000
Ozone (O ₃)	0-80	80-120	120-180	180-240	240-600
Sulphur dioxide (SO ₂)	0-100	100-200	200-350	350-500	500-1250

MONITORING NEEDED TO ADVISE PEOPLE!

e.g. qualar (from APA-Agência Portuguesa do Ambiente)

<https://qualar.apambiente.pt/qualar/index.php>

e.g. World Air quality index

<https://waqi.info/>



LIMITS!

PM 2.5 exposure of 25 $\mu\text{g}/\text{m}^3$



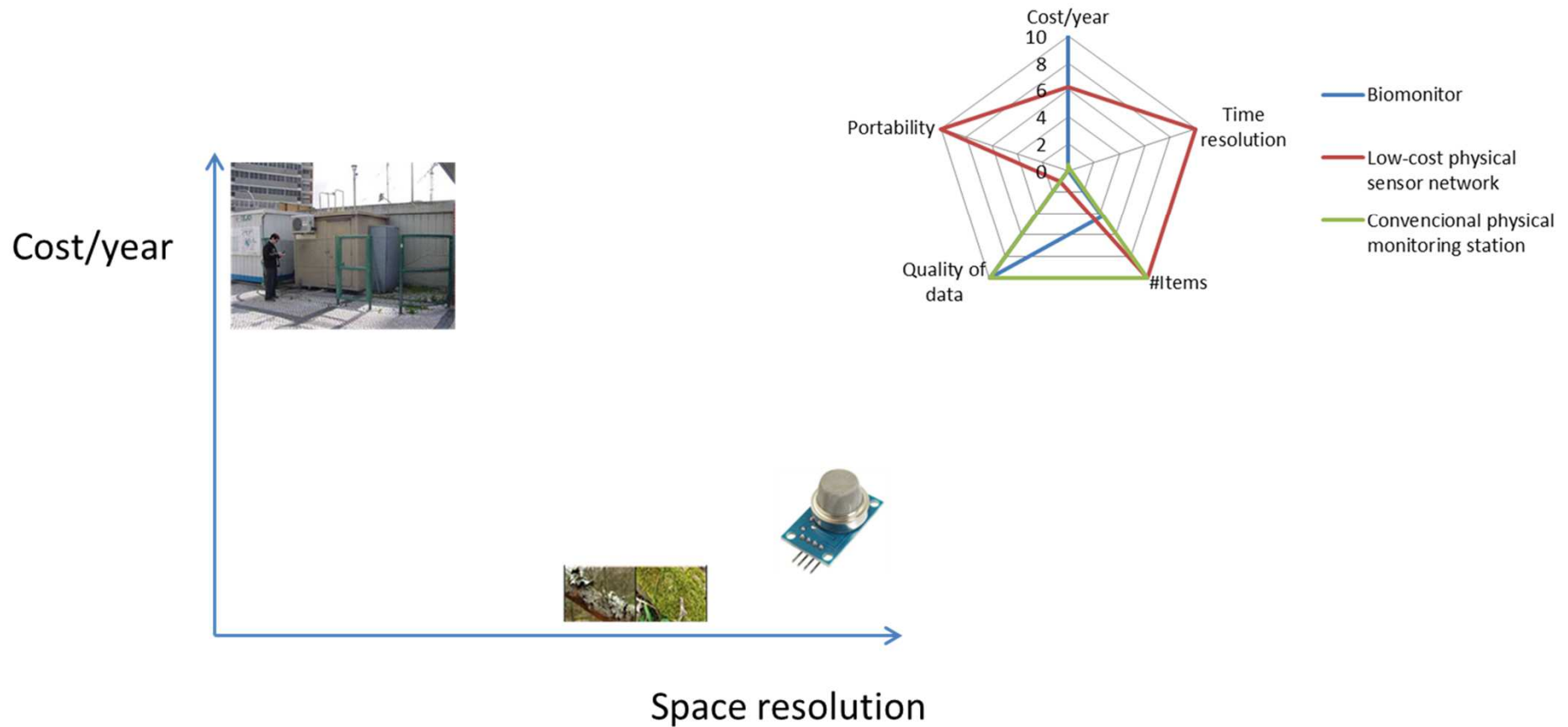
PM 2.5 exposure of 35 $\mu\text{g}/\text{m}^3$



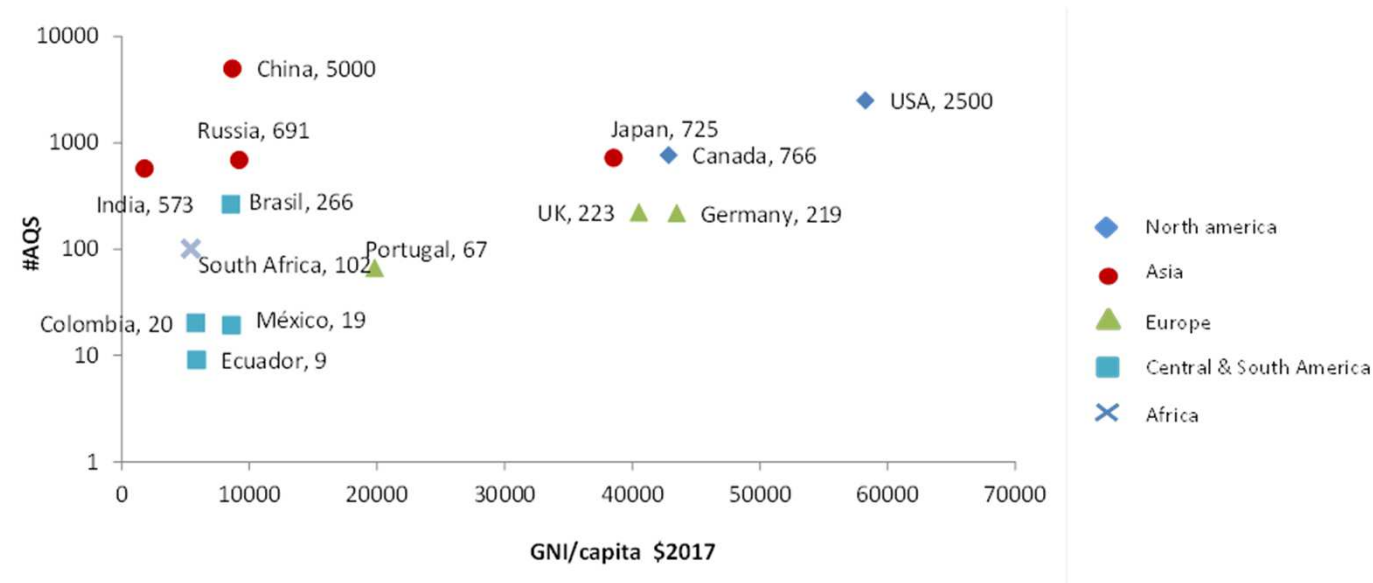
PM 2.5 exposure of 75 $\mu\text{g}/\text{m}^3$



MONITORING NEEDED TO ADVISE PEOPLE!



MONITORING NEEDED TO ADVISE PEOPLE!



AQS = number air quality monitoring stations

GNI = Gross National Income *

* is the total domestic and foreign output claimed by residents of a country, consisting of gross domestic product (GDP), plus factor incomes earned by foreign residents, minus income earned in the domestic economy by nonresidents

MONITORING NEEDED TO ADVISE PEOPLE!

Equipamento	Referência	Custo	Peso (kg)	âmbito	Portabilidade
	https://www.airpointer.com/	<i>High-Cost</i> 120 000€	80	Estação oficial compacta de qualidade do ar	
	https://www.qart.pt/solutions/qart-box/	<i>Medium-Cost</i> 4 000€	3 kg	Ruído, PM2.5, O3, NOx, SO2, H2S	
	http://aqicn.org/gaia/	<i>Low-Cost</i> 200 €	3 kg	PM2.5 e O3	

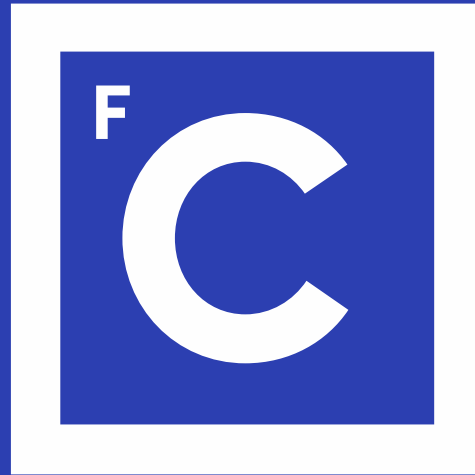
MONITORING NEEDED TO ADVISE PEOPLE!

	FCUL (Augusto, Máguas, Matos, Pereira, & Branquinho, 2010)	Low-Cost 500 €/análise	NA	Sensor biológico correlacionável com partículas, VOC, NOx	
	Rede de 6 sensores em escolas (Castell et al., 2018) da Aqmesh (https://www.aqmesh.com)	Low-Cost ~200 €	NA	CO, NO, NO₂ and O₃	
	Rede de 3 sensores numa estrada (Khedo, Perseedoss, & Mungur, 2010)	Low-Cost ~200 €	3 kg	CO, NO₂, PM2.5, O₃	

- Pollutants concentration and health;
- ppm conversion to g/m^3 ;
- Portuguese “Qualar” website; World air quality map;
- Air quality monitors high; medium and low-cost;
- “Qart box” example for air quality monitoring.

The image consists of a large blue square. Inside this square is a white border, which is itself a square. In the center of the white square is the word "Obrigado" in a bold, white, sans-serif font.

Obrigado



Ciências ULisboa

Faculdade
de Ciências
da Universidade
de Lisboa